

Deacetylation of α -Bromo- α -substituted-acetoacetates

Youhong Hu and Donglu Bai*

Shanghai Institute of Materia Medica, Chinese Academy of Science, 294 Taiyuan Road, Shanghai 200031, China

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Abstract: A new and efficient method for the deacetylation of α -bromo- α -substituted-acetoacetates to obtain α -bromoesters has been achieved by treatment with LiClO₄/Et₃N in ether at room temperature. © 1998 Elsevier Science Ltd. All rights reserved.

During the course of the study of [4+3] intramolecular cycloaddition, we used α -bromo- α -3-(2-furyl)propylacetoacetate(1) as one of the precursors. Under the conditions for the formation of allylic cation (Scheme 1), only deacetylated product 2 was obtained instead of the desired [4+3] cycloadduct 3. It is well-known that the hydrolysis of acetoacetates is readily accompanied with decarboxylation, but the efficient removal of the acetyl group from acetoacetates is rare. α -Bromoesters are versatile intermediates in organic synthesis for the preparation of phosphoranes and phosphonates in Wittig reaction or of α -amino acids and esters. This communication describes a facile method for the preparation of α -bromoesters by means of deacetylation of α -bromo- α -substituted-acetoacetates (Scheme 2).

Scheme 1

Methyl acetoacetate was converted to 5 in high yields by alkylation. The α-methine in 5 was brominated

by using MgBr₂ in the presence of H₂O₂, affording α -bromides 6 ³ in 72-82% yields. Deacetylation of 6 was achieved by treatment with LiClO₄/Et₃N in ether at room temperature for 4h to obtain α -bromoesters. The reaction sequence in Scheme 2 offers the possibility for preparation of α -bromoesters containing sensitive functionality in the substituent R to bromine and Ba(OH)₂.^{2a}

A typical procedure is as follows: Methyl α -bromo- α -3-(2-furyl)propylacetoacetate(1) (303mg, 1mmol) was dissolved in freshly distilled ether (20mL), LiClO₄ (203mg, 2mmol)and freshly distilled (from CaH₂) triethylamine (0.28mL, 2mmol) were added. The mixture was stirred at room temperature for 4h. The reaction was quenched with H₂O (5mL), and the mixture was extracted with Et₂O (15mL×3). The combined organic layer was washed with H₂O (10mL), brine (10mL), and dried over anhydrous Na₂SO₄. The solvent of the organic layer was removed in vacuum. The residual oil was purified by column chromatography on silica gel(petroleum ether : EtOAc = 20 : 1) to give compound 2 as a colorless oil (215mg) in 83% yield. In order to verify the generality of this reaction, the other five α -bromo- α -substituted-acetoacetates were tested and the yields of the deacetylated products 7 is in a range of 70 ~ 82% (Table 1). All the compounds obtained were characterized by ¹H NMR and IR spectra and new compounds were also characterized by MS, HRMS or elemental analysis.

Yield% Reaction Time Product 83 2* 4h 76 7a 6h CH3(CH2)2CH2-72 7b 5h 82 4h 2d 7d* 70

Table 1. Deacetylation of α -bromo- α -substituted-acetoacetate 6

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^{*} New compound